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A Dijodo-Substituted 9-Anthracenone

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Abstract

The central ring of the anthrone system in 1,8-diiodo-10-hydroxy-10-(3-methylbut-3-en-1-ynyl)anthracen-9(10H)-one, $C_{19}H_{12}I_2O_2$, has a boat conformation and the two outer rings form a dihedral angle of 41.1 (2)°. The I—C bond lengths are 2.094 (4) and 2.083 (4) Å. The hydroxyl and carbonyl groups form an intermolecular hydrogen bond having an $O \cdots O$ distance of 2.853 (4) Å.

Comment

The title compound, (I), was prepared as part of an ongoing project involving the structural study of key intermediates in the synthesis of substituted [3]-cumulenes (Garcia, Asfaw & Rodriguez, 1997), which have potential biological activity (Garcia, Ramos, Pratt & Rodriguez, 1995). The compound was prepared by a slight modification of a literature methodology (Garcia, Ramos, Pratt & Rodriguez, 1995) by the reaction of 1,8-diiodo-9,10-anthraquinone (Bayer, 1979) with the lithium salt of 2-methyl-1-buten-3-yne in tetrahydrofuran at 195 K.

Both the I1—C1 bond length of 2.094 (4) Å and the I2—C8 bond length of 2.083 (4) Å are comparable to the typical value of 2.095 Å for 51 measurements (Allen *et al.*, 1987). The central anthrone ring of the structure exists in a boat-like conformation having the torsion angles reported in Table 1. This conformation resembles that of 10-*tert*-butyl-9-anthracenone and almost all other 10-substituted 9-anthracenones (Sygula, Sygula, Fronczek & Rabideau, 1992). The two outer rings of

the structure are nearly planar, with the C1—C2— C3—C4—C13—C14 ring having a mean deviation of 0.0044 Å from planarity and the C5—C6—C7—C8— C11—C12 ring having a mean deviation of 0.0043 Å from planarity. The I1 atom lies 0.0772 (4) Å out of the plane of the ring to which it is bonded, while I2 lies a much greater distance of 0.1976 (4) Å out of the plane of its respective ring. The two outer rings form a dihedral angle of 41.1 (2)°. The intramolecular I1···O1 and I2...O1 distances are 3.171 (4) and 3.174 (4) Å, respectively, and are both shorter than the sum of the van der Waals radii (3.60 Å; Bondi, 1964). The I···O1···I angle is 101.4(1) Å. The intermolecular $O1 \cdot \cdot \cdot O2(\frac{3}{2} - x, 1 - y,$ $\frac{1}{2}$ + z) distance of 2.853 (4) Å is evidence of hydrogen bonding between the hydroxyl group and the ketone oxygen.

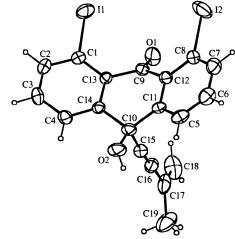


Fig. 1. The title molecule, with ellipsoids at the 40% probability level and H atoms represented with arbitrary radii.

Experimental

The title compound was prepared by a slight modification of the method of Garcia, Ramos, Pratt & Rodriguez (1995) by the reaction of 1,8-diiodo-9,10-anthraquinone (Bayer, 1979) with the lithium salt of 2-methyl-1-butene-3-yne in tetrahydrofuran at 195 K. The resulting yellow solid was recrystallized by slow evaporation from dichloromethane [m.p. 470 K (dec.)].

Crystal data

 $C_{19}H_{12}I_{2}O_{2}$ $M_{r} = 526.11$ Orthorhombic Pbca a = 15.117 (1) Å b = 14.4592 (6) Å c = 15.8539 (9) Å V = 3465.3 (6) Å³ Z = 8 $D_{x} = 2.017$ Mg m⁻³ D_{m} not measured Mo $K\alpha$ radiation $\lambda = 0.71073$ Å Cell parameters from 25 reflections $\theta = 8-14^{\circ}$ $\mu = 3.597$ mm⁻¹ T = 297 K Cuboidal $0.55 \times 0.50 \times 0.48$ mm Yellow

Data collection

Enraf-Nonius CAD-4 3815 reflections with diffractometer $I > \sigma(I)$ $\theta/2\theta$ scans $R_{\rm int} = 0.033$ $\theta_{\text{max}} = 29.97^{\circ}$ Absorption correction: $h = 0 \rightarrow 21$ ψ scans (North, Phillips $k = -20 \rightarrow 20$ & Mathews, 1968) $l = 0 \rightarrow 22$ $T_{\min} = 0.127, T_{\max} = 0.178$ 10 786 measured reflections 3 standard reflections 5037 independent reflections frequency: 120 min/ intensity decay: 0.82%

Refinement

-	
Refinement on F	$\Delta \rho_{\text{max}} = 1.36 \text{ e Å}^{-3}$
R = 0.045	$\Delta \rho_{\min} = -0.23 \text{ e Å}^{-3}$
wR = 0.054	Extinction correction:
S = 1.880	isotropic (Zachariasen,
3815 reflections	1963)
209 parameters	Extinction coefficient:
H atoms riding [C—H	$9.2(4) \times 10^{-8}$
$0.95 \text{ Å}; U(H) = 1.3 U_{eq}(C)$	Scattering factors from Inter-
$w = 4F_o^2/[\sigma^2(F_o^2)]$	national Tables for X-ray
$+ 0.0004F_a^4$	Crystallography (Vol. IV)
$(\Delta/\sigma)_{\text{max}} = 0.029$	

Table 1. Selected geometric parameters (Å, °)

IIC1	2.094 (4)	C10C14	1.534 (5)
I2C8	2.083 (4)	C10—C15	1.489 (6)
O1—C9	1.209 (4)	C15C16	1.185 (6)
O2C10	1.423 (4)	C16—C17	1.440(7)
C9C12	1.479 (5)	C17C18	1.345 (8)
C9C13	1.479 (5)	C17C19	1.486 (9)
C10C11	1.527 (6)		
I1C1C2	116.1 (3)	C11C10C14	110.8 (3)
I1C1C13	122.6 (3)	C11C10C15	108.1 (3)
12C8C7	116.4 (3)	C14—C10—C15	108.7 (3)
12C8C12	122.2(3)	C10C15C16	174.2 (4)
C12C9C13	115.3 (3)	C15C16C17	178.1 (5)
O2C10C11	110.4 (3)	C16C17C18	120.2 (5)
O2C10C14	107.0(3)	C16C17C19	115.5 (5)
O2C10C15	111.9 (3)	C18C17C19	124.3 (5)
C13-C9C12C11	36.3 (5)	C11C10C14C13	37.8 (5)
C12—C9—C13—C14	-37.4(5)	C10C11C12C9	2.0(5)
C14—C10—C11—C12	-37.8(5)	C9C13C14C10	-1.8(5)

The two largest peaks in the final difference map, both of density greater than 1 e Å^{-3} , were located less than 1 Å from the two I atoms.

Data collection: CAD-4 Operations Manual (Enraf-Nonius, 1977). Cell refinement: CAD-4 Operations Manual. Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: MULTAN80 (Main et al., 1980). Program(s) used to refine structure: MolEN. Molecular graphics: ORTEPII (Johnson, 1976). Software used to prepare material for publication: MolEN.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: CF1169). Services for accessing these data are described at the back of the journal.

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(1S,6S)-Tetrazolo[1,5-g]-7-aza-transhimachal-2-ene†

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Abstract

In the title compound, $C_{14}H_{22}N_4$, the tetrazole ring is fused with the eight-membered ring of the himachal-2-ene skeleton, which is in a *trans* conformation.

[†] Alternative name: (7aS,11aS)-4,5,6,7,7a,10,11,11a-octahydro-7,7,9-trimethyl-2-benzazocino[1,2-d]tetrazole.